

X-ray Absorption Spectra of Titanium-Containing Mesoporous SBA-15 Applied to CO₂ Photoreduction

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Photocatalytic reduction of CO₂ into high energy products has been extensively studied as a major alternative to address the greenhouse effect caused by excess CO₂ in the atmosphere, which also provides a non-fossil fuel. In this artificial photosynthesis process, TiO₂ is the mostly common used photocatalyst in aqueous and gaseous media due to its chemical stability, non-toxicity, and reasonably low cost. Hwang et al.¹ have demonstrated the utilization of mesoporous molecular sieves (MCM-41, MCM-48 and SBA-15) to highly disperse TiO₂ exhibits much higher photocatalytic reactivity for the reduction of CO₂ with H₂O as compared to bulk TiO₂, in which SBA-15 shows superior enhancement of photocatalytic reactivity. Thus, TiO₂ highly dispersed within mesoporous molecular sieves would facilitate the development of a process that could be designed for highly efficient photocatalytic systems.

In this work, titanium-containing mesoporous SBA-15 (Ti-vc-SBA (Ti/Si=0.01 (w/w))) was synthesized through a modified preparation process of SBA-15 by ternary surfactant system (C₁₆TMAB, SDS and P123), sodium silicate as Si source and (NH₄)₂TiF₆ as Ti source, under weak acidic conditions (pH=5), followed by grafting TiO₂ onto Ti-vc-SBA using TBOT/hexane solution in a cooling reflux system. Meanwhile, Ti-SBA synthesized via hydrothermal method using P123 alone as template was modified by TiO₂ (TiO₂/Ti-SBA) for comparison. Initially, the prepared materials were characterized by XRD, HRTEM, AAS, nitrogen sorption isotherm, UV-vis and XPS. X-ray absorption Near edge structure (XANES) data were collected using synchrotron radiations at NSRRC BL17C1 beam lines.

The dimensions of Ti-SBA are in the micrometer size range, whereas Ti-vc-SBA are quite thin (thickness < 100 nm). Most Ti-SBA have channels lying parallel to the surface, although some with perpendicular channels were observed. In contrast, the face-up nanochannels of Ti-vc-SBA were the only structures we saw. The XRD patterns of Ti-vc-SBA and Ti-SBA exhibit a peak of (1 0 0), which are the characteristic of a highly ordered hexagonal structure. UV-Vis spectra of TiO₂/Ti-vc-SBA and TiO₂/Ti-SBA show that the band edge of 360 nm is observed. The AAS and XPS data show the more amount of TiO₂ were added onto Ti-vc-SBA as compared to that of Ti-SBA.

A feature of the pre-edge spectrum at the Ti K-edge is sensitive to the symmetry of the surrounding atoms, being dipole forbidden, as indicated that the octahedral symmetry of Ti(IV) has a low intensity, whereas

tetrahedral symmetry is allowed and intensive². This effect has led to the extensive use of pre-edge spectra for establishing the coordination of titanium in oxide compounds. As shown in Fig. 1 (a) and (b), both Ti-vc-SBA and Ti-SBA exhibit an intense single pre-edge peak, indicating the titanium oxide species in both vc-SBA-15 and SBA-15 have tetrahedral symmetry of the titanium site. Highly dispersed titanium oxide species in the frameworks of SBA-15 were obtained. The presence of small, broad, additional peaks were detected as in Fig. (c) and (d), existing an aggregated octahedrally coordinated titanium oxide species as well as the tetrahedrally coordinated titanium oxide species. This implies extraframework titanium oxide species were formed after refluxing Ti-vc-SBA and Ti-SBA with TBOT/hexane solution. A correlation of the titanium content in the samples with XANES results relates the increase of the Ti/Si ratio with an increase of the highly coordinated titanium oxide species and a decrease of the pre-edge peak intensity. TiO₂/Ti-vc-SBA exhibited higher photocatalytic reactivity in CO₂ photoreduction than TiO₂/Ti-SBA, resulting from higher amount of highly dispersed titanium oxide species incorporated and higher accessible to reactants.

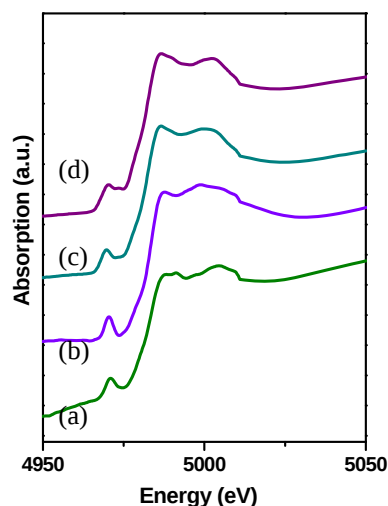


Figure 1. The XANES spectra of Ti-SBA (a), Ti-vc-SBA (b), TiO₂/Ti-SBA (c) and TiO₂/Ti-vc-SBA (d).

References:

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