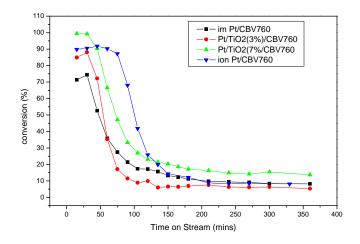
## Role of TiO<sub>2</sub> in Enhancing Sulfur Resistance of Supported Metal Catalysts (II): Correlation between Catalyst Structure and Catalytic Performance

Jyh-Fu Lee (李志甫)<sup>1</sup>, Hwo-Shuenn Sheu (許火順)<sup>1</sup>, Ku-Yin Lin (林谷音)<sup>2</sup>, and Jen-Ray Chang (張仁瑞)<sup>2</sup>

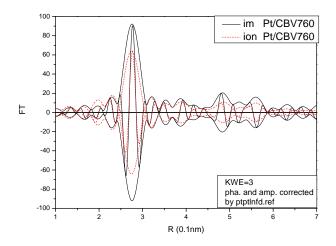
## <sup>1</sup>Department of Chemical Engineering, National Chung Cheng University, Chiayi, Taiwan <sup>2</sup>National Synchrotron Radiation Research Center, Hsinchu, Taiwan

The goal of this study is to improve the catalytic performance of a hydrogenation catalyst by retarding Pt aggregation during sulfur-poisoning catalyst deactivation. The improvement is motivated by the increase of metalsupport interactions. The increase of metal-support interactions was enhanced by grafting titanium oxide on zeolite surface. The efficacy of the modification was examined by aging tests of a hydrogenation reaction using 200 ppm sulfur-containing tetralin as a feed. The catalyst before and after the reaction were characterized by XAS (X-ray absorption spectroscopy), synchrotron XRD (X-ray diffraction). Correlated the structure characterization results with the catalytic performance tests, the following conclusions have been drawn:

- 1. The morphology of grafted TiO<sub>2</sub> is strongly dependent on TiO<sub>2</sub> loading. For the 3 Wt % TiO<sub>2</sub> loading, isolated TiO<sub>2</sub> has been observed. When the loading increases to 7 %, the TiO<sub>2</sub> morphology become monolayer while the loading further increase to 20 %, anatase TiO<sub>2</sub> appear.
- 2. The morphology of Pt clusters on  $TiO_2$ -zeolite is affected by  $TiO_2$  structure. Among the 3 samples, the Pt clusters on monolayer  $TiO_2$  presents the smallest particle size.
- 3. The initial activity of the catalyst sample is inversely proportional to particle size during the aging tests (Figure 1).
- 4. Small Pt clusters do not benefit in retarding Pt agglomeration during sulfur-poisoning catalyst deactivation. In stead, the increases of metal-support interactions do help the decrease of catalyst deactivation by the decrease of Pt agglomeration rate ( as shown in Figure 1 and 2 the sample prepared by ion-exchange having higher dispersion, whereas the conversion at steady state is almost the same ).
- 5. The role of the grafted  $TiO_2$  is regarded as an anchor that inhibits the Pt clusters from migration.



**Figure 1.** Comparison of the catalytic performance of the catalysts prepared by different methods, (operation conditions: Feed=200ppm sulfur containing tetralin, P=480psig,  $T=200^{\circ}C$ , WHSV=12,  $H_2$ /Oil mole ratio=7).



**Figure 2.** EXAFS characterizing Pt/CBV760 prepared from impregnation and ion-exchange technique. ( $k^3$  weighted, Pt-Pt phase and amplitude correction,  $3.5 < k < 14.0 \text{ Å}^{-1}$ )